

Cambridge International Examinations

Cambridge International General Certificate of Secondary Education

COMBINED SCIENCE 0653/51

Paper 5 Practical Test

October/November 2016

MARK SCHEME
Maximum Mark: 30

Published

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1 (a) cation

add sodium hydroxide solution/NaOH AND green ppt; (ALLOW gas changes red litmus to blue) iron(II)/Fe²⁺; anion add barium nitrate (solution)/Ba(NO₃)₂ **AND** white ppt; sulfate/SO₄²⁻; [4] (b) (i) colourless solution; fizzing / bubbles / effervescence; lighted splint pops; [4] hydrogen/H₂ (depends on use of splint); (ii) white ppt.; [2] (element L is) zinc/Zn; (a) time / minutes; [2] (beaker) A and (beaker) B/water and iodine solution; (b) (i) initial observation for **both** as cloudy/colourless/milky/white; [1] (ii) full set of results for both beakers; beaker A – no change to observation; beaker **B** – observation changes to blue-black/black; note: if all observations are black then (b)(i) not scored but award this ecf [3] (c) iodine (molecules) move into bag; by diffusion/because molecules are small enough; [3] starch and iodine produce blue-black/black/darker colour; (d) (blue-black) colour goes/colour fades/goes brown/goes orange/ goes yellow; [1]

2

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|---|---------|---|---------------|-----|
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| 3 | (a) (i) | sensible time = $31s \pm 0.5$ (accuracy mark); | | [1] |
| | (ii) | time recorded and less than that in (i); | | [1] |
| | (iii) | all time values recorded with pattern of decreasing values; | | [1] |
| | (b) (i) | T values calculated correctly (ignore no. of decimal places); | | [1] |
| | (ii) | complete set of T ² values recorded correctly rounded to 1 decimal | place ; | [1] |
| | (c) (i) | at least four plots correct to ½ small square ; | | |
| | | good best-fit line judgement ; | | [2] |
| | (ii) | (triangle method used) at least half of the line and indication on gra of how data obtained; | ph | |
| | | correct calculation using graph; | | [2] |
| | (d) 9.8 | ± 0.3 (accuracy mark) ; | | [1] |

Mark Scheme

Syllabus

Paper

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